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# IMPLICATIONS OF BUTYLATED HYDROXYTOLUENE AND OPTIMIZED PIPER NIGRUM EXTRACTS APPLICATION ON QUALITY OF STORED PORK

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# **Abstract**

Implications of butylated hydroxytoluene (BHT) and optimized Piper nigrum (White and Black Pepper) aqueous extracts (WPAE and BPAE) incorporation on fatty acid profile and physico-chemical indices of pork subjected to refrigeration storage was investigated. To produce optimized Piper nigrum aqueous extracts, desirability values of WPAE and BPAE were upscaled from 60 and 62% to 71 and 97% respectively using Design Expert, resulting in the yield of 65 and 87 bio-compounds. Fresh pork (100 g) was apportioned to four treatment groups (each replicated four times), comprising the Control (0% antioxidant); BHT (100 mg of BHT); BPAE (100 ml of BPE) and WPAE (100 ml of WPE), then soaked for 30 min prior to refrigeration storage [day (d) 0, 5 and 10] in a completely randomized design. All parameters examined were influenced (p<0.05) except additive effect on pork pH. Pork preserved with BPAE and WPAE minimized deterioration of fatty acids (FA) such as oleic, palmitoleic, total mono unsaturated fatty acid (MUFA) and poly unsaturated fatty acid (PUFA) than 0% antioxidant. BPAE had the highest MUFA: SFA, and alongside BHT, had significantly (p < 0.05) low index of atherogenicity. BPAE and WPAE had lesser cooking loss percentages than the Control. Additives X Storage day's effect reveal BPAE yielded least refrigeration loss (%) among samples evaluated and had extensive protective benefits. BPAE had the lowest milligram malondialdehyde per gram tissue on d 10 of storage. Extract from BP may potentially contain compounds that could offer enhanced meat preservation and promote consumer health.

Keywords: optimized Piper nigrum aqueous extracts, BHT, pig meat, fatty acid profile, oxidative stability, cooking and refrigeration losses, 2-thiobarbituric acid reactive substance

# INTRODUCTION

Nowadays, most consumers of meat are favorably disposed to meat products of higher nutritional value irrespective of the substance applied to limit deterioration or the storage employed, yet, healthconscious consumers are concerned of the role dietary sources play in the occurrence of diet-related diseases in humans (Beriain et al., 2018). Meat is a delicacy cherished for the peculiar sensations experienced by consumers from mastication as it

contributes to healthy diets and transfer of balancedprofile proteins. On worldwide basis, meat supply is projected to expand to meet consumer demand, expectedly reaching 377 Mt by 2031. Generally, pork consumption is projected to increase to 129 Mt in the next ten years (OECD/FAO, 2022). Consumers are attracted to pork due to its lower price and product consistency. In several tropical countries, favorable relative prices have positioned pork and poultry as choice meats to meet rising demand from the middle class. Aside its quality protein profile,









its B complex vitamins content such as vitamin B12 and mineral nutrients like zinc, selenium, iron and phosphorus makes it a desired choice (Pereira and Vincente, 2013; Bohrer, 2017). Pork, though commonly consumed contains less unsaturated fatty acid (UFA) compared to poultry. Muzolf-Panek, and Kaczmarek (2021) further explained that beef and pork contain significantly less poly unsaturated fatty acid (PUFA) and lower value for PUFA/UFA index compared to chicken meat.

Fatty acids (FA's) are important structural components of cells and biological membranes that are built into lipids derived components such as phospholipids, sphingolipids and lipoproteins. FA's contribute significant roles in fluid regulation in the human body by providing energy for membrane constituents of cells and influencing physiological activities in humans (Brown and Marnett, 2011; Dennis and Morris, 2015). FA's are likewise considered essential as cellular lipid mediators. In spite of numerous benefits that can be derived from FA's, oxidation of PUFA's during storage, processing and cooking significantly limit nutritional benefits humans derive. Additionally, food quality and shelf life are adversely impacted when unsaturated fats are oxidized. Reduced nutritional quality can translate into poor health for consumers, depending on the concentration of rancid substances in products. Fresh meat kept in the refrigerator ( $\leq 4$  °C) is prone to oxidative and microbial spoilage. Psychrotrophic facultative anaerobes can grow rapidly by depleting glucose and subsequently amino acids (Ling, 2015; Osman and Faruk, 2015; Wang et al., 2023) Quality preservation of meat therefore entails spoilage limitation by minimizing the extent of lipid oxidation and microbial proliferation.

Processing methods commonly applied during short term storage of meat include chilling, refrigeration and cooking. These methods are employed to conserve nutrients in meat as consumers are nourished with safe and quality products. While benefits of refrigeration and cooking are well documented, shrinkage of meat occurs, causing significant impact on meat quality and nutrient pool, though, the latter is notably critical in suppressing and destroying foodborne pathogens (Kondjoyan et al., 2014) Refrigeration and cooking significantly affect protein quality via structural and textural changes. The extent of evaporation of cellular moisture, rate of destruction of cell membranes, loss of cellular fibres and connective tissues as well as myofibrillar proteins contribute to the overall quality of stored meat (Pathare and Roskilly, 2016). Meat processors employ strategies to minimize Postmortem disruption of polypeptides in tissue though the disruption is essential for the development of flavour and eating qualities of meat. These disruptions initiates lipid oxidation, which unregulated, alters the nutritional, physico-chemical (colour and texture)

and sensorial properties (taste and aroma) of meat, causing off-flavour development that culminates into unacceptable taste and poor product quality (Lima *et al.*, 2013). Consequently, the search for safe substances and effective strategies to minimize loss of nutrients in meat is currently investigated.

There is an urgent need to develop safe and effective procedures to extract and isolate bioactive compounds since increased extraction efficiency results in the production of safe additives that are of significance to meat farmers and retailers. Biological activities of Piper nigrum (P. nigrum) seed and extracts have been explored and investigated. Piper species (white and black) are traditionally consumed to treat ailments or alternatively applied as antioxidant, preservative, insecticide or antimicrobial (Aziz et al., 2015; Gasparetto et al., 2017; Zhang et al., 2018). Extraction of organic compounds from very different types of matrixes using microwave technology has resulted in increased extraction efficiency and subsequent release of bio-active phytochemicals (Duarte et al., 2014). Currently, different kinds of solvents are used to access bio-active compounds in P. nigrum with varying yield and quality, however, the application of alcoholic extracts of plants directly on meat can result in significant shrinkage as well as global environmental implications; prompting the need for green methodology. The study, therefore, investigate the potency of butylated hydroxytoluene and P. nigrum extracts on quality of stored pork. It aims to prepare P. nigrum extracts using microwave heating in an environmentally friendly way, to compositionally identify bio-compounds in *P. nigrum* extracts using Gas Chromatography Mass Spectrometry (GC-MS), to describe the effect of prepared extracts on physico-chemical properties of pork during short term [24 h (d 0), 5 and 10 days] storage and to examine the potency of P. nigrum extracts on pork FA and lipid on d 10 of storage (4 °C).

# MATERIALS AND METHODS

# **Materials**

Fresh pork from the thigh (Biceps femoris) of the hind limb of a mixed breed pig of internal pH of 5.8 (24h postmortem) was sourced from a local market. Distilled water was obtained from the Laboratory of Animal Product and Processing, Department of Animal Production and Health. Butylated hydroxytoluene [2,6-Di-tert-Butyl-p-Cresol (99% purity) by Loba Chemie] was gotten from a reputable store. Dried peppercorns of black and white pepper were sourced from a reputable local herb market, separately grinded using an attrition mill (industrial grinding mill Gx200 petrol engine, India). Afterwards, powdered peppers were clarified using an 0.40 mm sieve (Adegoke et al., 2013; Adegoke et al. submitted), then sealed and stored in air-tight amber colored vials.

# Generation of Extraction Criteria for the Production of Piper Nigrum Aqueous Extracts

Extraction parameters were generated by modifying production constraints from data provided [Adegoke et al. (2023); submitted data] by Design Expert Software. Post-upscale of previous data, 97% desirability was achieved after setting the following targets. Extraction time was set to range between 85 and 95 min; solvent volume was fixed between 360 and 380 mL; extract volume was set between 60.42 and 115.15 mL; extract absorbed was defined as maximized between 3 and 7 mL; pH on d 0 and 5 were targeted at 5.85 and 6.05 respectively; pH on d 10 was set to range between 6.37 and 7; meat 2-thiobarbituric acid reactive substance (TBARS) value on d 5 and 10 was targeted at 0.3 and 0.11 mg malondialdehyde (MDA) g<sup>-1</sup> tissue respectively, while cooking loss was set to range between 30.6 and 41%.

To achieve 71% desirability for white pepper extract, Design Expert software was employed. Extraction time was fixed between 85.2 and 95 min; solvent volume was fixed to range between 280 and 295 mL; extract volume was set as minimized at 41 mL; extract absorbed was set as maximized between 0.4 and 6.3 mL; pH on d 0 and 5 were defined as targeted at 5.61 and 6.05 respectively; pH on d 10 was set as minimized at 6.6; and 2-thiobarbituric acid reactive substance (TBARS) values on d 5 and 10 were set as minimized, and targeted at 0.277 and 0.230 mg (MDA) g-1 tissue respectively; while cooking loss was fixed at 17.54%. Afterwards, extraction conditions in Tab. I were proffered. Subsequently, ground *P. nigrum* powders were screened with a 0.40 mm sieve. Thereafter, eight grams was separately weighed per pepper and dissolved in 404.21 and 414.21 mL of distilled water for white and black pepper respectively. Each mixture was thoroughly stirred in microwavetolerant containers and placed in heating cavities of Samsung (GE109MST) and Daewood (KOR6N9NC) microwave ovens respectively, then subsequent heating at 300 and 320 W was performed for 85.27 and 86.91 min to produce white and black pepper aqueous extracts (WPAE and BPAE). Afterwards, containers were unloaded and allowed to cool prior to evaluation of bio-compounds extracted via Gas Chromatography Mass Spectrometry (GC-MS) analysis.

# Analysis of Bio-Compounds in Optimized Aqueous Extracts

Spectrometry (GCMS analysis) of optimized P. nigrum extracts generated was carried out. An Agilent Technologies Network GC system was set up comprising GC (Agilent 7890A, USA) and MS (Agilent 5975C model, USA) with column model Agilent technologies HP5MS of 30 m length, 0.32 mm internal diameter and 0.25  $\mu$ L thickness. Agilent Technologies GC injector of initial temperature of 80 °C was held for 2 min at 12 degrees per min, and

I: Criteria for desirability values prior to production of white and black pepper aqueous extracts

Number	Pepper extract	Ext Time (min)	Solvent vol (mL)	Extract	Extract Absorbed	Meat pH d 0	Meat pH d 5	Meat pH d 10	TBARS d 5	TBARS d 10	Cook loss % I	Desirability
1	White	85.273	280.000	65.997	5.698	6.284	6.057	6.587	0.275	0.308	18.997	0.713
2	Black	806.98	364.206	95.229	7.014	5.850	6.081	6.887	0.301	0.137	35.620	0.971
Ext – extractio	n; TBARs – 2-t	xt – extraction; TBARs – 2-thiobarbituric acid value (MDA);	cid value (MD		d – day; Ref – Refrigeration; Desirability was upscale from Adegoke <i>et al.</i> (2023); Adegoke <i>et al.</i> (submitted)	n; Desirabilit	y was upscale	from Adegoke	et al. (2023); A	degoke et al.	(submitted)	

raised to 240 degrees for 6 min. Little portion of the supernatant solution was sieved through a  $0.45\,\mu m$  micro filter and acetone of analytical grade was added at 1:10. Subsequently, a microliter of solution was injected into the GC according to procedure reported by Adegoke *et al.* (2023). Afterwards, spitless mode of analysis was carried out using scan ranges between 50 and 500, followed by identification of compounds according to the apex fragmentation fingerprints area for extracts of black (retention time of  $4.56-20.40\,min$ ) and white pepper (retention time of  $3.33-20.40\,min$ ).

### **Experimental Treatment Groups**

Two repetitions (microwave extraction) were performed to constitute each P. nigrum group. A total of four experimental groups, each consisting four replicates were formed. Treatments comprised the Control (100 g of Pork + 0% antioxidant + 1 g of NaCl + 5 mL of Olive oil + 100 mL of distilled water), BHT [100 g of Pork + 0.1 g of BHT (Naveena et al., 2008) + 1 g of NaCl + 5 mL of Olive oil + 100 mL of distilled water], WPAE [100 g of pork + 5 mL of olive oil + 1 g of NaCl + 100 mL of white pepper extract (WPE)] and BPAE [100 g of pork + 5 mL of Olive oil + 1 g of NaCl + 100 mL of black pepper extract (BPE)] groups outlaid in a Completely Randomized Design. All compositions were thoroughly mixed prior to pork insertion. Olive oil was added to aid the dissolution of BHT, while its inclusion across all replicates was carried out to maintain uniformity. Next, one hundred grams (100 g) of pork apportioned to each replicate was completely immersed for 30 min. Afterwards, immersed samples were removed and placed in soft disposable plastic trays wrapped with packaging film and placed in refrigerator in preparation for subsequent quality evaluations across storage (4°C) days.

#### **Data Collection**

#### Meat Fatty Acid Profile

On d 10 of storage (±4°C), lipid was extracted from pork according to method established by Folch et al. (1957). In total, 10 g of pork was cut out, blended and homogenized. Chloroform: methanol (2:1, v/v) mixture was added to blended meat at 40 mL: 2 g. Samples were incorporated at 1 µL in split mode (20:1). Separation was performed using chromatograph equipped with a flame ionization detector and GC column [HP-88 (Agilent Technologies, USA)  $(30 \text{ m} \times 0.25 \mu\text{m} \times 0.25 \text{ mm})$ BPX-70 fused silica capillary column)]. Initial column temperature was held at 115 °C for 2 min at 10 °C/min after which temperature was increased to 200 °C held at a rate of 18.5 min at 60 °C min<sup>-1</sup>, and raised to 245 °C for 4 min. Injector and detector temperature was 250 °C. Helium gas was passed through the column at 1.0 flow rate, 14.6 pressure and velocity of 29. Injector temperature was held

at 300 °C, while hydrogen flow occurred at 40.0 but air flow at the detector was 184.0. Further flow of helium gas occurred at 45.0 through a temperature of 300 °C. Absorbance of concentration of different standard solutions specific to fatty acid and sample benzene extracts were measured using a spectrophotometer according to wavelength identified for each fatty acid. Area under each peak was identified by fatty acid methyl esters that were compared with standard AOCS (1998) Ce1f-96. Result was presented in mg/100 g of tissue. Relationship between the addition of the main proatherogenic saturated and antiatherogenic unsaturated fatty acids were computed to obtain pork index of atherogenicity as reported by Ulbricht and Southgate (1991):

$$IA = \frac{(C12:0 + (4 X C14:0) + C16:0)}{\sum MUFA + \sum (n-6) + \sum (n-3)}.$$

### Meat pH

To calibrate the pH meter (model pH – 108A by ATC Hanna, India), the probe of the pH meter was inserted into buffer 7, quickly cleaned, then inserted into buffer 4. Afterwards, the electrode of the standardized hand-held pH meter was inserted deeply into pork containing additive to obtain the initial pH. The procedure was repeated on day (d) 5 and 10.

# Meat Refrigeration Loss

Samples containing additives were re-weighed after 24 h (d 0), d 5 and 10 to determine weight change, which was subsequently expressed as a percentage of the initial weight.

Refrigeration weight loss was calculated as follows:

Refrigeration loss (%) =

= weight before refrigeration – weight after refrigeration weight before refrigeration

# **Meat Cooking Loss**

After determining refrigeration loss percentage on d 10, samples were weighed, wrapped in separate air-tight polythene bag and cooked in a digital water bath (Uniscope Sm801a, Surgifriend Medicals, England) at 70 °C for 30 minutes (Sanwo *et al.*, 2019). Cooking loss percentage was calculated as follows:

Cooking loss (%) =

 $= \frac{\text{Weight before cooking - Weight after cooking}}{\text{weight before cooking}} \times 100.$ 

# Meat Malondialdehyde Content

On d 10 of refrigeration storage, ten grams  $(10\,g)$  of pork sample was cut out per replicate

and blended with 20 mL extracting of solution (7.5% trichloroacetic acid + 0.1% ethylenediamenetetraacetic acid + 0.1% propyl gallate) for 30 s using a tissue homogenizer. Afterwards, mixture was decanted for 1 min and filtrate (using muslin cloth) was centrifuged [Merlin 503 by Spectra Scientific Limited, UK) at 2000 rpm for 15 min. Next, 1 mL of supernatant extract was added to 1 mL of TBA reagent in labelled test tubes. The test tubes (larger) were slightly sealed (to allow air escape) and placed in water bath [Uniscope Laboratory Water Bath (Model SM801A) by Surgifriend Medicals, England] at 70°C for 40 min, then cooled with running water. Aliquots were afterwards transferred to 96 well plates and absorbance of pink chromogen formed against blank was measured at 532 nm using a UV-VIS spectrophotometer (Helios, Thermo-spectronic, Cambridge, UK). Results were presented as malondialdehyde equivalents (mg MDA/g tissue of

# Statistical Analysis

Data obtained from physico-chemical and fatty acid profile assessments were subjected to Oneway analysis of variance (ANOVA) in a Completely Randomized Design. Data for additives × storage days was outlaid in a 4 × 3 factorial arrangement with additives and storage days as fixed effects and the replicates as a random effect; and analysed using the General Linear Mixed Model (GLMM) of Statistical Package for Social Sciences (SPSS) version 26 (2019). Significant means at 5% level of significance were separated using Duncan Multiple Range Test (DMRT) of the same Statistical Package.

#### RESULTS

GC-MS analysis for optimized WPAE and BPAE yielded 65 and 87 bio-compounds respectively (Tab. II and III). WPAE contained 22 nitrogen-based compounds, consisting of one fluorine-nitrogen based compound, one bromine-nitrogen based compound, four nitrogen-sulphur based compounds

II: Gas chromatography mass spectrometry of bio-compounds in white pepper aqueous extract

S/N	Compound	Molecular formula	Area
1	3-Ethoxy-1,1,1,5,5,5-hexamethyl-3(trimethylsiloxy)trisiloxane	$C_{11}H_{32}O_4Si_4$	1.03
2	Cyclohexene, 4-methyl-1-(1-methylethenyl)	$C_{10}H_{18}$	1.06
3	2-Methyl-6-(5-methyl-2-thiazolin-2-ylamino) pyridine	$C_{10}H_{13}N_3S$	1.06
4	Benzaldehyde, 2-methoxy-	$\mathrm{CH_3OC_6H_4CHO}$	1.06
5	1-Piperidinecarboxaldehyde	$C_6H_{11}NO$	1.09
6	1-(2-Adamantylidene) semicarbazide	$C_{11}H_{17}N_3O$	1.13
7	5,6-Epoxy-2,2-dimethyl-3-heptyne	$C_9H_{14}O$	1.13
8	2,4-Decadienamide, N-isobutyl-, (E, E,)	$C_{14}H_{25}NO$	1.18
9	8-Oxabicyclo [3.2.1] oct-6-en-3-one, 2,4-dimethyl	$C_9H_{12}O_2$	1.18
10	1,1,3,3,5,5,7,7-Octamethyl-7-(2-methylpropoxy) tetrasiloxan-1-ol	$C_{12H_{34}O_5Si_4}$	1.21
11	Hexasiloxane, tetradecamethyl-	$C_{14}H_{42}O_5Si_6$	1.22
12	1-(2-Adamantylidene) semicarbazide	$C_{11}H_{17}N_3O$	1.31
14	Cyclopentasiloxane, decamethyl-	$C_{10}H_{30}O_{5}Si_{5}$	1.32
15	2-Amino-4-methylphenol, 2TMS derivative	$\mathrm{C_{13}H_{25}NOSi_2}$	1.32
16	Salicylic acid, 2TMS derivative	$C_{13}H_{22}O_3Si$	1.32
17	Cyclohexasiloxane, dodecamethyl	$C_{12}H_{36}O_6Si_6$	1.39
18	6-Anilino-2-(m-tolyl)-1,3,5-thiadiazine-4-thione	$\mathrm{C_3H_2N_2S_2}$	1.41
19	Ethylene glycol, 2-methoxybenzoate	$C_{10}H_{12}O_{2}$	1.41
20	Benzene, 2,4-diethyl-1-methyl-	$C_{11}H_{16}$	1.53
21	Pyrido[2,3-d] pyrimidine, 4-phenyl-	$C_{13}H_{9}N_{3}$	1.53
22	N-Methyl-1-adamantaneacetamide	$C_{13}H_{21}NO$	1.53
23	Caryophyllenyl alcohol	$C_{15}H_{26}O$	1.55
24	Thiophene, 2,3,4-trimethyl-	$C_7H_{10}S$	1.55
25	2-Acetylcyclopentanone	$C_6H_{10}O$	1.55

S/N	Compound	Molecular formula	Area
26	(-)-Spathulenol	$C_{15}H_{24}O$	1.55
27	(3-Methylphenyl) phenylmethanol	$C_{14}H_{14}O$	1.55
28	2-Methyl-4-(2,6,6-trimethylcyclohex-1-enyl) but-2-en-1-ol	$C_{14}H_{24}O$	1.55
29	Trimethylsilyl [2-(4-chlorophenyl)-4-phenyl-1,3-thiazol-5-yl] acetate	$\mathrm{C_{25}H_{24}ClNO_{2}SSi}$	1.68
30	2,5-Dihydroxybenzoic acid, 3TMS derivative	$C_{16H_{30}O_{4}S}$	1.68
31	Trisiloxane, 1,1,1,5,5,5-hexamethyl-3,3-bis[(trimethylsilyl)oxy]	$C_{12}H_{36}O_4Si$	1.68
32	3-Trimethylsilyloxystearic acid, trimethylsilyl ester	$C_{24}H_{52}O_3Si_2$	1.94
33	1,1,1,5,7,7,7-Heptamethyl-3,3-bistrimethylsiloxy) tetrasiloxane	$C_{13}H_{39}O_{5}Si_{6}$	1.94
34	Cyclotetrasiloxane, octamethyl-	$C_8H_{24}O_6Si_5$	2.04
35	4-Ethylbenzamide	$C_9H_{11}NO$	2.06
36	Phthalic acid, isobutyl 2-methoxyethyl ester	$C_{15}H_{20}O_{5}$	2.06
37	$1 \ H-Cycloprop[e] azulen-7-ol, decahydro-1,1,7-trimethyl-4-methylene-\ [1ar-(1a.\ alpha.,4a.\ alpha.,7.beta.,7a.beta.,7b.alpha.)]$	$C_{15}H_{24}O$	2.24
38	(3aS,4R,7R)-1,4,9,9-Tetramethyl-5, 6,7,8-tetrahydro-4H-3a,7-methanoazulene	$C_{15}H_{22}$	2.24
39	[3-(4-Methoxyphenyl)-4,5-dihydro-1,2-oxazol-5-yl] methanol	$C_{10}H_{13}NO_3$	2.24
40	2-Ethylacridine	$C_{15}H_{13}N$	2.24
41	Tris(tert-butyldimethylsilyloxy)arsane	$\mathrm{C_{18}H_{45}AsO_{3}Si_{3}}$	2.24
42	Benzamide, N-(1,1-dimethylethyl)-4-methoxy	$C_{12}H_{18}N_2O_2$	2.53
43	1,2-Benzisothiazole-3-propanoic acid	$C_{10}H_9NO_2S$	2.53
44	(E)-5-(Benzo[d][1,3] dioxol-5-yl)-1-(piperidin-1-yl)pent-2-en-1-one	$\mathrm{C_{17}H_{21}NO_{3}}$	2.53
45	Pentasiloxane, dodecamethyl-	$C_{12}H_{36}O_4Si_5$	2.59
46	Octasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11,13,13,15,15-hexadecamethyl	$C_{16H_{48}O_{7}Si_{8}}$	2.59
47	3-Isopropoxy-1,1,1,7,7,7-hexamethy l-3,5,5-tris(trimethylsiloxy)tetrasiloxane	$C_{18H_{52}O_{7}Si_{7}}$	2.59
48	l- (+)-Lactic acid, tert-butyldimethylsilyl ether	$C_{15H_{34}O_{3}Si_{2}}$	3.05
49	Allyl(2-butoxy) dimethylsilane	$C_9H_{20}OSi$	3.05
50	Chloromethyl cyanide	C <sub>2</sub> H <sub>2</sub> ClN	3.05
51	Benzene, 1,2-dichloro-	$C_6H_4Cl$	3.99
52	Benzene, 1,3-dichloro-	$C_6H_4Cl_2$	3.99
53	4-Methylnaphtho[1,2-b] thiophene	$C_{13}H_{10}S$	5.77
54	Benzene, 1-fluoro-4-(2-phenylethenyl)	$C_{14}H_9F$	5.77
55	Tacrine	$C_{13}H_{14}N_2$	5.77
56	Furane-2-carbohydrazide, 5-phenylethynyl	$C_{13}H_{10}N_2O_2$	7.83
57	Butylphosphonic acid, butyl 5-methoxy-3-methylpentyl ester	$C_{14}H_{31}O_4P$	7.83
58	1,2-Diborane (4) diamine, 1,2-diethyl-N, N, N', N'-tetramethyl	$C_8 H_{22} B_2 N_2$	7.83
59	Caryophyllene	$C_{15}H_{24}$	8.68
60	Tetrasiloxane, decamethyl-	$\mathrm{C_{10}H_{30}O_{3}Si_{4}}$	10.01
61	Arsenous acid, tris(trimethylsilyl) ester	$C_9H_{27}AsO_3Si_3$	11.80
62	Cyclotrisiloxane, hexamethyl-	$C_6H_{18}O_3Si_3$	11.80
63	Oxime-, methoxy-phenyl-	$C_8H_9NO_2$	13.73
64	Silane, dimethyl (dimethyl (but-2-en yolky)silyloxy) (but-2-enyloxy)	$\mathrm{C_{12}H_{26}O_{2}Si}$	13.73
65	Oxazolidine, 3-phenyl-	C <sub>9</sub> H <sub>11</sub> NO	13.73

III: Gas Chromatography Mass Spectrometry of bio-compounds in black pepper aqueous extract

S/N	Compound	Molecular formula	Area
1	Ethyl 4-hydroxymandelate, 2TMS derivative	$C_{16H_{28}O_{4}Si_{2}}$	0.49
2	3-Hydroxymandelic acid, ethyl ester, di-TMS	$\mathrm{C_{16}H_{28}O_{4}Si_{2}}$	0.49
3	p-Trimethylsilyloxyphenyl-bis(trimethylsilyloxy)ethane	$C_{17H_{34}O_{3}Si_{3}}$	0.49
4	4-(4-Methylphenylamino)pyrido [3,2-c] pyridazine	$C_{14}H_{12}N_4$	0.49
5	3-Amino-7-nitro-1,2,4-benzotriazine 1-oxide	$C_7H_5N_5O_3$	0.49
6	trans-3-Ethoxy-b-methyl-b-nitrostyrene	$C_{11}H_{13}NO_3$	0.49
7	1-Piperidinecarboxaldehyde	$C_6H_{11}NO$	0.54
8	3,3-Dimethylpiperidine	$C_7H_{15}N$	0.54
9	Caprolactam	$C_6H_{11}NO$	0.54
10	Cyclopentasiloxane, decamethyl-	$C_{10}H_{30}O_{5}Si_{5}$	0.59
11	N-(Trifluoracetyl)-O,O',O''-tris (trimethylsilyl)epinephrine	$C_{20}H_{36}F_3NO_4Si_3$	0.59
12	$N\hbox{-(Trifluoroacetyl)-N,O,O',O''-tetrakis(trimethylsilyl)}\ no repine phrine$	$C_{22}H_{42}F_3NO_4Si_4$	0.59
13	Bicyclo[3.1.1]hept-2-ene, 3,6,6-trimethyl	$C_{10}H_{16}$	0.60
14	1,3,7-Octatriene, 3,7-dimethyl-	$C_{10}H_{16}$	0.60
15	. beta. –Ocimene	$C_{10}H_{16}$	0.60
16	2-Propenoic acid, 2-methyl-3-(4-nitrophenyl)	$C_{10}H_9NO_4$	0.60
17	Piperonal	$C_8H_6O_3$	0.63
18	Heptasiloxane,1,1,3,3,5,5,7,7,9,9,11,11,13,13-tetradecamethyl	$C_{14}H_{42}O_6Si_7$	0.63
19	3-Amino-2-phenazinol ditms	$C_{18}H_{25}N_3OSi_2$	0.63
20	Ar-tumerone	$C_{15}H_{20}O$	0.67
21	3-Methyl-2-butenoic acid, 3-phenyl-2-propenyl Ester	$C_{14}H_{18}O_2$	0.67
22	(Z, Z) alpha. –Farnesene	$C_{15}H_{24}$	0.73
23	1-(2-Adamantylidene) semicarbazide	C <sub>11</sub> H <sub>17</sub> N <sub>3</sub> O	0.73
24	1,7-Octadiene, 3,6-dimethylene-	$C_{10}H_{14}$	0.73
25	[1,2,4] Triazolo[1,5-a]pyrimidine-6-carboxylic acid, 4,7-dihydro-7-imino-, ethyl ester	C <sub>10</sub> H <sub>12</sub> N <sub>4</sub> O <sub>3</sub> S	0.77
26	Pyrrolo[2,3-g]indazole-7,8-dione, 1,6-dihydro-	$C_9H_5N_3O_2$	0.77
27	Camphene	$C_{10}H_{16}$	0.77
28	Diazoprogesterone	$C_{21}H_{30}N_4$	0.79
29	Bicyclo[3.1.0]hexane, 6-methylene-	C <sub>7</sub> H <sub>10</sub>	0.79
30	1H-Pyrazole-1-acetamide,4-iodo-N-(phenylmethyl)	$C_{12}H_{12}IN_3O$	0.90
31	gamma. –Neoclovene	C <sub>15</sub> H <sub>24</sub>	0.90
32	1H-3a,7-Methanoazulene,octahydro-1H-3a,7-Methanoazulene, octahydro-	$C_{16}H_{26}O_{2}$	0.90
33	Phenol, 2,6-dimethoxy-	$C_8H_{10}O_3$	1.05
34	Phenol, 3,4-dimethoxy-	$C_8H_{10}O_3$	1.05
35	3-Amino-2,6-dimethoxypyridine	$C_7H_{10}N_2$	1.05
36	1-((1S,3aR,4R,7S,7aS)-4-Hydroxy-7-isopropyl-4-methyloctahydro-1H-ind en-1-yl) ethenone	C <sub>15</sub> H <sub>26</sub> O <sub>2</sub>	1.13
37	((1R,4S,5R)-1-Methyl-4-(prop-1-en-2yl) spiro[4.5]dec-7-en-8-yl) methanol	C <sub>15</sub> H <sub>24</sub> O	1.13
38	Phenol, 4-methyl-2-nitro-	C <sub>7</sub> H <sub>7</sub> NO <sub>3</sub>	1.13
39	Cyclopentylsilane	C <sub>5</sub> H <sub>9</sub> Si	1.20
40	Cyclohexanol, 4-methoxy-	$C_7H_{12}O_2$	1.20
41	3-Octadecanone	C <sub>18</sub> H <sub>36</sub> O	1.20

S/N	Compound	Molecular formula	Area
42	2-Fluoro-4-methylpyridine	$C_6H_6FN$	1.35
43	2-Acetylcyclopentanone	$\mathrm{C_7H_{10}O_2}$	1.35
44	1,1,1,5,7,7,7-Heptamethyl-3,3-bis(trimethylsiloxy)tetrasiloxane	$C_{13}H_{39}O_{5}Si_{6}$	1.47
45	Cyclohexasiloxane, dodecamethyl-	$C_{12}H_{36}O_{6}Si_{6}$	1.47
46	Oxazepam, 2TMS derivative	$C_{15}H_{11}ClN_2O_2$	1.47
47	6,7-Dimethyl-1,2,3,5,8,8a-hexahydronaphthalene	$C_{12}H_{18}$	1.51
48	Sospathulenol	$C_{15}H_{24}O$	1.51
49	Benzene, 1-(1,5-dimethyl-4-hexenyl	$C_{15}H_{22}$	1.59
50	Cyclohexene,3-(1,5-dimethyl-4-hexenyl)-6-methylene-, [S- (R*, S*)]	$C_{15}H_{24}$	1.59
51	Cedrene	$C_{15}H_{24}$	1.59
52	gamma. – himachalane	$C_{15}H_{24}$	1.59
53	2,5-Dihydroxybenzoic acid, 3TMS derivative	$C_{16}H_{30}O_4Si$	1.88
54	Cyclotetrasiloxane, octamethyl-	$\mathrm{C_8H_{24}O_6Si_5}$	1.93
55	2-Methoxy-4-vinylphenol	$C_{9}H_{10}O_{2}$	1.98
56	4-Hydroxy-3-methylacetophenone	$C_9H_{10}O_2$	1.98
57	Cycloheptasiloxane, tetradecamethyl	$C_{14}H_{42}O_{7}Si_{7}$	2.38
58	Trisiloxane,1,1,1,5,5,5-hexamethyl-3,3-bis[(trimethylsilyl)oxy]	$C_{12}H_{36}O_{4}Si$	2.38
59	Pentasiloxane, dodecamethyl-	$C_{12}H_{36}O_4Si_5$	2.38
60	8-Methyl-4-azafluorenone	$C_{19}H_{14}N_{2}$	2.46
61	2-Amino-9-fluorenone	$C_{13}H_{9}NO$	2.46
62	4-Amino-9-fluorenone	$C_{13}H_{9}NO$	2,46
63	2-Bromo-2'-methoxyacetophenone	$C_9H_9BrO_2$	2.66
64	Ethylene glycol, 2-methoxybenzoate	$C_{10}H_{12}O_3$	2.66
65	2-Methoxy-N'-[4-(trifluoromethyl)pyridin-2-yl]benzohydrazide	C <sub>7</sub> H <sub>6</sub> F <sub>3</sub> NO	2.66
66	2-Phenyl-3-(2-furyl)-propenal	$C_{13}H_{10}O_{2}$	2.77
67	2-furancarboxaldehyde, 5-[2-phenylethenyl]	$C_{13}H_{10}O_{2}$	2.77
68	Phosphonous acid, phenyl-, diethylester	$C_{10}H_{15}O_{2}P$	2.77
69	2-Naphthalenecarboxylicacid,8-ethenyl-3,4,4a,5,6,7,8,8a-octahydro5-methylene	$C_{18}H_{22}O_{2}$	3.20
70	Caryophyllene oxide	$C_{15}H_{24}O$	3.20
71	(-)-Spathulenol	$C_{15}H_{24}O$	3.20
72	6,6-Dimethyl-2-vinylidenebicyclo [3.1.1] heptane	$C_{11}H_{16}$	3.50
73	Caryophyllene	$C_{15}H_{24}$	3.50
74	trans beta. –Ocimene	$C_{10}H_{16}$	3.50
75	Dibenzothiophene, 4-methyl-	$C_{13}H_{10}S$	4.31
76	Benzene 1-fluoro-3-(2-phenylethen yl)- (E)		4.31
77	Dibenzothiophene, 3-methyl-	$C_{13}H_{10}S$	4.31
78	Hexane, 1-chloro-5-methyl-	$C_7H_{15}Cl$	5.03
79	(3R,4aS,8aS)-8a-Methyl-5-methylene-3-(prop-1-en-2-yl)-1,2,3,4,4a,5,6,8a-octahydronaphthalene)	$C_{15}H_{22}$	5.03
80	6,7-Dimethyl-1,2,3,5,8,8a-hexahydr 1] hept-2-ene	$C_{10}H_{18}$	5.03
81	Tris(tert-butyldimethylsilyloxy)arsane	C <sub>18</sub> H <sub>45</sub> AsO <sub>3</sub> Si <sub>3</sub>	10.28
82	Arsenous acid, tris (trimethylsilyl) ester	$C_9H_{27}AsO_3Si_3$	10.28
83	Cyclotrisiloxane, hexamethyl-	C <sub>6</sub> H <sub>18</sub> O <sub>3</sub> Si <sub>3</sub>	10.83

S/N	Compound	Molecular formula	Area
84	1-Chloromethyl-4-(1,1-diethylpropy l) benzene	$C_{15}H_{23}Cl$	10.90
85	Thiophene, 2-heptyl-5-propyl-	$C_{14}H_{24}S$	10.90
86	Furane-2-carbohydrazide, 5-phenylethynyl	$C_{13}H_{10}N_2O_2$	10.90
87	Piperine	$C_{17}H_{19}NO_{3}$	24.11

IV: Main effect of additives and storage days on physio-chemical parameters of stored pork

Donomotono			Additive				Storag	ge days	
Parameters	Control	BHT	BPAE	WPAE	SEM	0	5	10	SEM
рН	5.95	6.00	5.92	5.87	0.06	5.01 <sup>c</sup>	6.30 <sup>b</sup>	6.50a	0.05
REF loss (%)	8.39b	$9.14^{\rm b}$	7.72 <sup>b</sup>	12.71a	0.51	3.39 <sup>c</sup>	9.83 <sup>b</sup>	15.24ª	0.44
Cooking loss (%)	39.77ª	36.99 <sup>ab</sup>	36.34 <sup>b</sup>	36.06b	0.60	0	0	0	0

a,b,c – means on the same row with different superscript differ significantly (p < 0.05)

BHT – Butylated hydroxytoluene; WPAE- white pepper aqueous extract; BPAE- black pepper aqueous extract; REF – refrigeration; SEM – standard error of mean

V: Interactive effect of additive and storage days on pH and refrigeration loss of stored pork

Additive		Control	l		BHT			BPAE			WPAE		
Day	0	5	10	0	5	10	0	5	10	0	5	10	
рН	5.22c	6.24 <sup>b</sup>	6.40ab	5.31 <sup>c</sup>	6.31 <sup>ab</sup>	6.41 <sup>ab</sup>	4.82 <sup>d</sup>	6.39ab	6.57ab	4.71 <sup>d</sup>	6.26b	6.64ª	0.11
REF Loss (%)	2.34 <sup>h</sup>	$9.34^{\rm ef}$	13.50 <sup>bc</sup>	2.68 <sup>h</sup>	9.65 <sup>def</sup>	15.08b	2.46 <sup>h</sup>	8.61 <sup>fg</sup>	12.10 <sup>cd</sup>	6.11 <sup>g</sup>	11.73 <sup>cde</sup>	20.28a	0.80

 $^{a,b,c,d,e,f,g,h}$  – means on the same row with different superscript are significantly (p < 0.05) different BHT – Butylated hydroxytoluene; WPAE – white pepper aqueous extract; BPAE – black pepper aqueous extract; REF – refrigeration; SEM – standard error of mean

and one nitrogen-boron-based compound. It also contains silicon-based compounds (22) and sulphur containing compounds (6) that comprise one sulphur-silicon based compound. Four chlorinebased compounds comprising 2 chlorine-nitrogen based compounds were found. Arsane containing compounds (2) and phosphorus-based compound (1) were likewise extracted. Area of yield ranged from 1.03–13.73% with 3-Ethoxy-1,1,1,5,5,5-hexamethyl-3(trimethylsiloxy)trisiloxane least and Oxime-, methoxy-phenyl-, Silane, dimethyl (dimethyl (but-2-en yolky) silyloxy) (but-2-enyloxy) and Oxazolidine, 3-phenyl- highest. For BPAE, twentyfive nitrogen-containing compounds were present, with chlorine-nitrogen (1), nitrogen-sulphur (1), fluorine-nitrogen (1), iodine-nitrogen (1) and fluorine-nitrogen-sulphur-chlorine-based compound (1) found. Six sulphur-based compounds were gotten as well as three chlorine-containing compounds. One phosphorus-based and brominebased compound were respectively seen. A total of nineteen silicon-based compounds were harvested, while two arsane-containing compounds were extracted. Piperine was the most abundant biocompound harvested, spanning an area of 24.11%. Compositional analysis of extracts revealed rich diversity of element-based compounds. This is a significant upgrade to outcomes reported by Olalere *et al.* (2018) and Adegoke *et al.* (2023) that had 31 and 71 bio-compounds respectively.

Main effect of storage days and additives (0% antioxidant, BHT, and *P. nigrum* extracts) on pork pH, refrigeration (REF) loss and cooking loss is shown (Tab. IV). Sole effect of additives and storage days was significant (p < 0.05), except pH. Pork pH was numerically highest among meat samples containing BHT but lowest among WPAE samples. Weight loss (%) from refrigeration was increased when WPAE was added to pork than 0% antioxidant, BHT and BPAE additives. Cooking loss (%) of pork containing *P. nigrum* extracts was reduced compared to 0% antioxidant group.

From parameters assessed (Tab. IV), both pH and refrigeration loss (%) were affected (p < 0.05) exclusively by refrigeration storage days. The same statistical significance was identified for meat pH and refrigeration loss percentage. As days of refrigeration storage extended, the pH and weight loss increased (p < 0.05), with highest loss recorded on d 10.

The interaction effect of additives and storage days on pork pH and refrigeration loss is documented (Tab. V). All parameters examined were influenced (p < 0.05). On d 0, both *P. nigrum* groups had lower pH than Control and BHT. Peak pH was recorded for WPAE on d 10. Refrigeration loss (%) increased across storage days and treatment. Loss (%) from

VI: Influence of aqueous additives on fatty acid profile pork on d 10 of storage (4 °C)

Parameters (mg/100 g tissue)	Control	BHT	BPAE	WPAE	SEM
Lauric (12:0)	2.65b	1.98°	2.73 <sup>b</sup>	3.15 <sup>a</sup>	0.11
Stearic (18:0)	$7.70^{b}$	6.80°	7.85 <sup>b</sup>	8.28 <sup>a</sup>	0.15
Palmitic (16:0)	$3.33^{b}$	2.53°	3.50 <sup>ab</sup>	$3.74^{a}$	0.13
Myristic (14:0)	$0.90^{a}$	0.28 <sup>c</sup>	0.21 <sup>c</sup>	$0.48^{b}$	0.07
Butyric (4:0)	1.18a	0.73b	1.18ª	1.33ª	0.07
Caprylic (8:0)	1.85 <sup>b</sup>	1.48°	1.90 <sup>b</sup>	2.28ª	0.08
Valeric (5:0)	1.30 <sup>ab</sup>	0.85°	1.15 <sup>b</sup>	1.55ª	0.08
SFA	18.89 <sup>b</sup>	14.65°	18.47 <sup>b</sup>	20.85ª	0.62
Oleic (18:1; ω-9)	3.73 <sup>c</sup>	2.83 <sup>d</sup>	4.08b	4.40a	0.16
Palmitoleic (16:1; ω-7)	$0.35^{b}$	$0.50^{\rm ab}$	0.70ª	0.60ª	0.05
MUFA	$4.08^{b}$	3.33°	4.78ª	5.00 <sup>a</sup>	0.18
MUFA: SFA	1.76 <sup>c</sup>	$1.85^{bc}$	2.09ª	1.93 <sup>b</sup>	0.04
Arachidonic (20:4; ω-6)	3.95 <sup>b</sup>	4.60a	3.95 <sup>b</sup>	4.53ª	0.10
Linoleic (18:2; ∞-6)	$3.63^{b}$	2.88 <sup>c</sup>	4.13ª	4.35 <sup>a</sup>	0.15
PUFA	7.58 <sup>c</sup>	7.48°	8.08b	8.88ª	0.35
PUFA: SFA	4.01 <sup>c</sup>	5.11ª	4.37b	$4.26^{\mathrm{bc}}$	0.12
UFA	11.70°	10.80 <sup>d</sup>	12.90 <sup>b</sup>	13.90ª	0.10
UFA: SFA	6.19 <sup>d</sup>	7.37ª	5.63b	5.37°	0.11
IA	8.68ª	5.40b	5.72 <sup>b</sup>	7.27 <sup>ab</sup>	0.47

a, b, c, d – means on the same row with different superscript differ significantly (p < 0.05)

BHT – Butylated hydroxytoluene; BPAE – Black Pepper Aqueous Extract; WPAE – White Pepper Aqueous Extract; MUFA – Monounsaturated fatty acid; SFA – Saturated fatty acid; UFA – Unsaturated fatty acid; PUFA – Polyunsaturated fatty acid; SEM – standard error of mean; IA – index of atherogenicity

WPAE samples was highest across treatment groups on d 0, but lesser in BPAE on d 5 and 10 though similar (p < 0.05) as the Control.

Fatty acid profile of pork containing aqueous additives is presented in Tab. VI. Saturated fatty acid (SFA) in meat include lauric, stearic, palmitic, butyric, myristic, caprylic, valeric fatty acid. Monounsaturated and polyunsaturated fatty acids (MUFA and PUFA) present were oleic and palmitoleic; linoleic and arachidonic acid respectively. All fatty acid indices measured were influenced (p < 0.05), but pork N3: N6 ratio was not affected. An identical statistical significance was observed treatment groups for lauric and stearic acid, with WPAE highest and BHT lowest. The control group contained more myristic acid than BHT and BPAEpreserved pork. WPAE contained more palmitic acid than BHT and Control, though similar as BPAE. Butyric acid was higher among Control, WPAE and BPAE than BHT. Identical superscript was assigned to caprylic and total SF acid values. Pork containing WPAE had more SF acids than Control and BPAE; while both latter groups had more than BHT. BHT samples had less valeric acid than WPAE. Also, oleic acid was reduced in WPAE, BPAE, Control and BHT respectively (p < 0.05), while palmitoleic acid was high in WPAE and BPAE than the Control. Identical

superscript was assigned linoleic acid and total MUFA. Pork containing BHT had lesser MUFA than the Control and P. nigrum groups. Also, MUFA: SFA ratio was reduced (p < 0.05) in the Control compared to BPAE that was highest. Arachidonic acid was higher in BHT and WPAE samples than BPAE and the Control, while linoleic fatty acid was reduced in BHT samples than the Control, but highest among WPAE and BPAE groups. Pork PUFA was more in meat containing WPAE than BPAE, however, both Control and BHT had identical total PUFA values. though least. Pork PUFA: SFA proportion was least for the Control but highest in BHT. Statistically identical superscripts were recorded for oleic acid and UFA, with BPAE highest but BHT lowest. Pork UFA: SFA was least in the Control, followed by WPAE, then BPAE but BHT was highest. Meat index of atherogenicity (IA) was low in BHT and BPAE compared to the Control but WPAE was intermediate (p < 0.05).

The 2-thiobarbituric acid reactive substance (TBARs) of pork soaked in aqueous additives stored under refrigeration condition for 10 days is revealed (Tab. VII). Pork TBARs was significantly (p < 0.05) affected. WPAE had the highest TBARs, followed by the Control, but BPAE was lowest.

VII: Oxidative status of pork containing aqueous additives on d 10 of refrigeration storage

Parameter	Control	BHT	BPAE	WPAE	SEM
TBARS (mg MDA/g tissue)	5.385 <sup>b</sup>	4.930°	3.455 <sup>d</sup>	7.015ª	0.328

 $^{a,\,b,\,c,\,d}$  – means on the same row with different superscript differ significantly (p < 0.05)

BHT – Butylated hydroxytoluene; WPAE – white pepper aqueous extract; BPAE – black pepper aqueous extract; MDA – malondialdehyde; TBARS – 2 thiobarbituric acid reactive substance; SEM – standard error of mean

#### DISCUSSION

Increase in meat pH across storage (4°C) days, solely due to refrigeration was influenced by end products released from the metabolism of materials from protein origin alongside protein degradation facilitated by the proliferation of psychrotrophic microorganisms (Rey et al., 1976). Biogenic amines formed from amino acid decarboxylase action of microbes elevate pH during storage (4°C) by complementing contributory activity of peptides, dipeptides, and free amino acids produced. This process in turn changes the structural composition of pork due to changes in myofibrils and sarcolemma linkages. Once degraded, the connection between myofibrils and the cell membrane is lost, thus creating a free-rein of water movement and subsequent water loss from the cells of stored meat (Straadt et al., 2007).

Lower refrigeration loss in BPAE may be attributed to the combined activity of phosphate and dissolved salt (NaCl). Puolanne and Ruusunen (1983) and Offer and Knight (1988) explained that phosphate markedly enhance water-holding properties of NaCl at 1.0 to 1.5% as a result of the cleavage of actin and myosin bonds within the muscle by phosphates, thus facilitating the enlargement of filament lattice within meat (Warner, 2017). The water holding capacity of pork muscle was affected, evidenced by a decline in pH among BPAE samples. Also, 1-Chloromethyl-4-(1,1-diethylpropy l) benzene, Hexane, 1-chloro-2-Bromo-2'-methoxyacetophenone, 1H-Pyrazole-1-acetamide,4-iodo-N-(phenylmethyl), N-(Trifluoracetyl)-O,O',O"-tris (trimethylsilyl) BPAE alongside compounds such as Chloromethyl 1,2-dichloro-; cyanide, Benzene, Benzene. 1,3-dichloro-; Benzene, 1-fluoro-4-(2-phenylethenyl), 4-Methylnaphtho[1,2-b] thiophene, 2-Methyl-6-(5-methyl-2-thiazolin-2-ylamino) pyridine play supportive role by restricting the permeability of muscle cells against higher permeability of muscle membrane.

Pork endo-, peri- and epimysial collagens are generally affected after cooking, causing shortening of actin and myosin filament, reduced muscle cell diameter and increased release of juice in meat that comprise of fat and water. In the control group, degradation of protein subunits results in disruption of myofibril as less pork juice is entrapped within protein structures by capillary forces. Reduced water holding capacity of pork in the control group increased the cooking loss of meat (Lorenzo *et al.*, 2015; Botinestean *et al.*, 2018).

Rapid decline in pH of pork from combined activity of additive and refrigeration storage on d 0 among WPAE and BPAE samples may be attributed to rise in hydrogen ion concentration, primarily as a result of dissociation of weak acids. Some phytochemicals such as 3-Hydroxymandelic acid, ethyl ester, di-TMS; 2-Propenoic acid, 2-methyl-3-(4nitrophenyl); 3-Methyl-2-butenoic acid, 3-phenyl-2-propenylEster, Arsenous acid, tris (trimethylsilyl) ester; 2-Naphthalenecarboxylicacid,8-ethenyl-3,4,4a, 5,6,7,8,8a-octahydro5-methylene and Phosphonous acid, phenyl-, diethylester in BPAE; as well as Salicylic acid, 2TMS derivative; 2,5-Dihydroxybenzoic acid, derivative; Phthalic acid, isobutyl 2-methoxyethyl ester; 3-Trimethylsilyloxystearic acid, trimethylsilyl ester and 1,2-Benzisothiazole-3-propanoic acid in WPAE are weak organic acids that can tilt meat pH towards acidity. Munteanu and Apetrei (2021) explained that the extent of hydrogen ion donation by phytochemicals depends on several factors such as the chemical structure. concentration and pH of the solution. Hence, hydrogen-rich phytochemicals can act as proton donors (donate hydrogen ions), lowering pork pH by increasing the concentration of hydrogen ions (H+). Also, subsequent rise in pH among WPAE and BPAE preserved pork like other groups on d 5 and 10 suggests loss in strength of acids as storage days extends. Jokanović and Cvetković (2009) identified oxidation of weak acids and anaerobic degradation of organic acids as possible causes of rise in pH as storage extends; and on d 10, all samples were affected in similar ways by storage conditions. On d 10, refrigeration loss was lowered among BPAE samples than other groups possibly due to the rich matrix of organic acid present. While the strength of organic acids lowered during storage likely from dilution effect of hygroscopic pull of moisture available from salts present, it could however be deduced that the power of concentration influences to a larger degree the quality of produce stored. The diminishing influence of organic bio-compounds is similar to loss of strength of ascorbic acid during storage (Lisiewska et al., 2003). Additionally, the presence of fluorine-nitrogen based compound, bromine-nitrogen based compound, nitrogensulphur based compound, sulphur containing compounds and abundantly rich hydrogen-based compounds likely modified the initial pH of pork, causing a decline, though not significant. This may have resulted in significant subtle shrinkage of the myosin head, causing changes in myofibrillar space and subsequent weight during refrigeration. Meat with low pH, increased reduction in intra-fibrillar space corresponds with increased loss of attraction between the protein and water molecules. This is commonly induced in meat marinated in acidic conditions (Huff-Lonergan and Lonergan, 2005).

Reports have detailed the significance of fatty acid to man's diet. Mensink (2013) explained that low SFA in the diet lowers blood total serum and LDL cholesterol concentrations. Nowadays, generally acceptable dietary compositions tend towards lowering SFA in man's diet while increasing the UFA content. The result obtained tends to favour the application of BHT on pork, however, careful assessment of the interrelationship between the SFA and total MUFA and PUFA for BHT may reveal some likely implications on cardiovascular health of consumers. In this study, pork MUFA and PUFA proportions were conserved by the radical quenching or/and antimicrobial potency of organic additives. Though WPAE had more influence on conservation of oleic fatty acid, similar amounts of palmitoleic and total MUFA as BPAE was documented. Pork soaked in WPAE and BPAE with higher proportions of UFA suggests that the mode of action of bio-compounds extends beyond its antioxidant protection. Microbes deteriorate fatty acids, especially lipolitic bacteria, which left uncontrolled, could catalyze hydrolysis of fat, resulting in the production of aldehydes and acids. Microorganisms in the genera of Alteromonas, Achromobacter, Flavobacterium, Pseudomonas, Staphylococcus and Yeast are lipolytic. Research has shown that lipolytic microbes can hydrolyze triglycerides from extracellular lipases produced using lipases like triacylglycerol and acyl hydrolase enzymes to speed up the hydrolysis process (Chandra et al., 2020). Piperine – the principal bioactive abundant in BPAE is capable of altering the permeability of bacterial cell wall composed of high level of lipid material. Wenxue et al. (2018) examined the influence of black pepper on Staphylococcus aureus and E coli species and reported that 1H-Cycloprop[e]azulen-7-ol, decahydro-1,1,7-trimethyl-4-methylene-.  $(1a\alpha,4a\alpha,7\beta,7a,\beta,7\beta\alpha.)$ ] - found in the chloroform extract of black pepper degraded bacterial cell walls and membranes alongside the inhibition of protein synthesis. Similarly, 1H-Cycloprop[e]azulen-7-ol, decahydro-1,1,7-trimethyl-4-methylene- [1ar-(1a. alpha.,4a. alpha.,7. beta., 7a. beta.,7b. alpha.)] in WPAE possibly stifled microbial growth by the release of intracellular transaminases; causing degradation of cell wall and membranes of fatdecomposing lipolytic bacterial species. Higher proportion of MUFA and MUFA: SFA, especially among BPAE samples, followed by WPAE is desirable as diets with favorable cardiovascular protection and an improved lipid profile is sought after by meat consumers (Leah et al., 2011). MUFA: SFA proportion is more important in assessing

the quality of dietary lipids consumed than the PUFA: SFA ratio, which is substantiated by the recommendation of World Health Organization that balanced proportions of dietary fats (SFA: MUFA: PUFA) at 1: 1.5: 1 is recommended (Grover *et al.*, 2021). From this recommendation, BPAE and WPAE samples appear appealing than others.

From the perspective of PUFA/SFA index commonly used to examine the role diet plays on cardiovascular health; BHT-incorporated pork appears healthier for human consumption since a higher ratio corresponds to healthier influence, but as Chen and Liu (2020) explained, not all of the main classes of PUFA positively contribute to the prevention of arterial or cardiac diseases, and not all SFA's elevate serum cholesterol. The findings of Dietschy (1998) showed that C12:0 (lauric), C14:0 (stearic), and C16:0 (palmitic) may contribute to the elevation of serum cholesterol concentration by limiting the activity of low-density lipoprotein receptors but C18:1 n-9 cis (oleic acid), increases the activity of low-density lipoprotein receptors (LDLRs) and decreases the cholesterol concentration in serum. Maragoni et al. (2020) points to dietary intake of linoleic acid (LA, C18:2 n-6) as inversely correlated with arterial or cardiac diseases. Thus, the proportion of anti-cardiac to pro cardiac ratio for BHT, BPAE and WPAE according to Chen and Liu (2020) is implicative. Anti-cardiac [(C18:1+ C18:2) BHT samples (0.571), BPAE samples (0.821) and WPAE samples (0.875)] to pro-cardiac [(C12:0+C14:0+C16:0) BHT (0.479), BPAE (0.644) and WPAE (0.737)] ratio equals 1.19: 1.27: 1.19. This implies that consuming pork stored with WPAE may result in low exposure to arterial plaque development similar as BHT-incorporated meat, though without any safety concern or risk to health, yet, better than both, BPAE samples potentially least exposes consumers to the development of arterial plaque.

Least TBARs in BPAE may be due to rich phytochemicals of biological significance in black pepper extract. (-)-Spathulenol in BPAE is a sesquiterpenoids alkaloid with antioxidant capacity. Similarly, Cyclotrisiloxane, hexamethylis a major bio-compound found in seaweed with antioxidant properties (Adedapo et al., 2011). Other compounds such as. beta. -Ocimene, Ar-tumerone, Caryophyllene, trans-. beta. –Ocimene and piperine are few radical quenching compounds from the abundant pool of phytochemicals extracted. Radical quenching phytochemicals (phenols, alkaloids, sesquiterpenoids, terpenes and other compounds) in medicinal plants exhibit redox quenching properties by adsorbing and neutralizing free radicals including singlet and triplet oxygen or by decomposing peroxide in cell, tissue and meat (Adedapo et al., 2011; Moyo et al., 2022). The absence of piperine in WPAE likely limited the potent expression of WPAE as an anti-oxidant on d 10 of storage.

#### **CONCLUSION**

Pork containing BPAE is desirable for its minimal oxidative rancidity and lower moisture loss – a significance to meat consumers and processors. While BPAE and WPAE had higher and highest UFA respectively, this study has showed that extract from BP may potentially contain compounds that could offer enhanced meat preservation as shown by MUFA: SFA and TBARs values for BPAE-preserved pork. Though this study investigated the benefits of aqueous extracts up till d 10, a more potent means for limiting deterioration may be explored such as a combination of updated storage technologies alongside application of optimized extract.

#### Disclosure statement

The authors report there are no competing interests to declare.

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